

**School of Chemistry  
Faculty of Sciences  
University of the Punjab, Lahore  
Course Outline**



**Semester-VII**

Programme	BS Chemistry	Course Code	Chem-461	Credit Hours	2
Course Title	Fuel Chemistry & Renewable Energy Resources	Course Type	Major (Elective)		
Course Introduction					
The students will learn about the existing energy resources; their processing, refining and utilization. Processing of crude oil: Brief description of origin of petroleum, Petroleum processing like Desalting, Fractional distillation; refining, cracking; reforming, isomerization; alkylation and finishing processes (Doctor's Sweetening and Merox processes) Coal chemicals & Renewable energy resources: The destructive distillation of coal, coking of coal distillation of coal tar; Liquid Fuel and Hydrogenolysis, Solar Energy and Solar Cells, Wind and Water Energy Geothermal Energy and Biofuels					
Learning Outcomes					
On the completion of the course: <ul style="list-style-type: none"><li>Students are expected to become familiarized with the concepts of Fuel chemistry</li><li>This will enable them qualify for basic to moderate level jobs involving general knowledge of chemistry</li><li>The obtained knowledge shall also enable the students to enter into various entrepreneurial activities involving general introduction to chemistry.</li><li>Students are able to understand the concept of GLP and GMP</li></ul>					
Course Content				Assignments/Readings	
Week 1	Brief description of origin of petroleum			Class Based learning/tests	
Week 2	Desalting of petroleum			Class Based learning/tests	
Week 3	Fractional Distillation of Petroleum			Class Based learning/tests	
Week 4	Fractional Vacuum Distillation of Petroleum			Class Based learning/tests	
Week 5	Cracking of Petroleum			Class Based learning/tests	
Week 6	Reforming of Petroleum			Class Based learning/tests	
Week 7	Isomerization of Petroleum			Written Assignment	
Week 8	Alkylation of Petroleum			Class Based learning/tests	
Week 9	Midterm Assessment			----	
Week 10	Finishing Processes (Doctor's Sweetening and Merox processes)			Class Based learning/tests	
Week 11	Destructive distillation of coal, coking of coal			Class Based learning/tests	

<b>Week 12</b>	Distillation of Coal Tar	Class Based learning/tests	
<b>Week 13</b>	Liquid Fuel and Hydrogenolysis	Class Based learning/tests	
<b>Week 14</b>	Solar Energy and Solar Cells	Class Based learning/tests	
<b>Week 15</b>	Wind and Water Energy	Quiz	
<b>Week 16</b>	Geothermal Energy and Biofuels	Class Based learning/tests	
<b>Textbooks and Reading Material</b>			
1. Applied Chemistry, Haq Nawaz Bhatti and Muhammad Salman, 2017, Caravan Book Publisher, Pakistan. 2. Water Supply and Sewerage, T.J.McGhee, McGraw Hill Book Co. New York.(1991) 3. Hand Book of Industrial Chemicals, By SIRI Board of Consultants and Engineers, 4. Shereve’s Chemical Process Industries, 5th Ed.1975 by G.T.Austin McGraw Hill Book Co. New York. 5. Industrial chemistry, B. K. Sharma Krishna Prakashan Media (P) Ltd., Ed-15 (2006) 6. Environmental Chemistry by Stanely E. Manahann			
<b>Teaching Learning Strategies</b>			
1. Lectures 2. Group Discussion 3. Laboratory work 4. Seminar/ Workshop			
<b>Assignments: Types and Number with Calendar</b>			
1.Written 7 <sup>th</sup> week 2. Quiz 15 <sup>th</sup> week			
<b>Assessment</b>			
<b>Sr. No.</b>	<b>Elements</b>	<b>Weightage</b>	<b>Details</b>
1.	Midterm Assessment	35%	Written Assessment at the mid-point of the semester.
2.	Formative Assessment	25%	Continuous assessment includes: Classroom participation, assignments, presentations, viva voce, attitude and behavior, hands-on-activities, short tests, projects, practical, reflections, readings, quizzes etc.
3.	Final Assessment	40%	Written Examination at the end of the semester. It is mostly in the form of a test, but owing to the nature of the course the teacher may assess their students based on term paper, research proposal development, field work and report writing etc.

Semester VII					
Programme	BS Chemistry	Course Code	Chem-462	Credit Hour	1
Course Title	Applied Chemistry Lab-I		Course Type	Major (Elective)	
Course Introduction					
It will increase knowledge in the practical handling of fuels and their characterization. Also knowledge about the heavy metal load in industrial effluents will be acquired. Determination of Diesel index, Aniline point and pour point of petroleum products, Proximate analysis of Coal, Analysis of Industrial Effluents (Chromium, Iron, Chlorides etc.)					
Learning Outcomes					
On the completion of the course: <ul style="list-style-type: none"><li>Students are expected to become familiarized with the concepts of general chemistry</li><li>This will enable them qualify for basic to moderate level jobs involving general knowledge of chemistry</li><li>The obtained knowledge shall also enable the students to enter into various entrepreneurial activities involving general introduction to chemistry</li><li>Students are able to understand the concept of GLP and GMP</li></ul>					
Course Content			Assignments/Readings		
Week 1	Determination of Diesel Index of Diesel Oil		Lab work / Notebook		
Week 2	Determination of Aniline Point of Diesel oil		Lab work / Notebook		
Week 3	Determination of Cetane Number of Diesel oil		Lab work / Notebook		
Week 4	Determination of Pour Point of Fresh Mobil Oil		Lab work / Notebook		
Week 5	Determination of Pour point of Used Mobil Oil		Lab work / Notebook		
Week 6	FTIR Assay of used and fresh Mobil Oil		Lab work / Notebook		
Week 7	Determination of Smoke Point of Fresh Mobil Oil		Written Assignment		
Week 8	Determination of Smoke Point of Used Mobil Oil		Lab work / Notebook		
Week 9	Mid Term Examination		Lab work / Notebook		
Week 10	Analysis of Industrial Effluents (Chromium)		Lab work / Notebook		
Week 11	Analysis of Industrial Effluents (Chromium)		Lab work / Notebook		
Week 12	Analysis of Industrial Effluents (Iron)		Lab work / Notebook		
Week 13	Analysis of Industrial Effluents (Iron)		Lab work / Notebook		
Week 14	Analysis of Industrial Effluents (Chlorides)		Lab work / Notebook		
Week 15	Analysis of Industrial Effluents (Chlorides)		Quiz		
Week 16	Review and Viva Voce		Lab work / Notebook		

Textbooks and Reading Material			
1. Applied Chemistry, Haq Nawaz Bhatti and Muhammad Salman, 2017, Caravan Book Publisher, Pakistan. 2. Petroleum Refining Technology, Ram Parsad (2002). 3. Industrial chemistry, B. K. Sharma, Krishna Prakashan Media (P) Ltd., Ed-15 (2006). 4. Shereve's Chemical Process Industries, 5th Ed.1975, by G.T.Austin, McGraw Hill Book Co. New York.			
Teaching Learning Strategies			
1. Lectures 2. Group Discussion 3. Laboratory work 4. Seminar/ Workshop			
Assignments: Types and Number with Calendar			
1. Written 7 <sup>th</sup> week 2. Quiz 15 <sup>th</sup> week			
Assessment			
Sr. No.	Elements	Weightage	Details
1.	Midterm Assessment	35%	Written Assessment at the mid-point of the semester.
2.	Formative Assessment	25%	Continuous assessment includes: Classroom participation, assignments, presentations, viva voce, attitude and behavior, hands-on-activities, short tests, projects, practical, reflections, readings, quizzes etc.
3.	Final Assessment	40%	Written Examination at the end of the semester. It is mostly in the form of a test, but owing to the nature of the course the teacher may assess their students based on term paper, research proposal development, field work and report writing etc.

Semester VII					
Programme	BS Chemistry	Course Code	Chem-463	Credit Hours	2
Course Title	Steel Industry and Metal Finishing	Course Title	Major (Elective)		
Course Introduction					
<p>The students will learn about the metallurgical operation regarding steel industry as well as classical and advance technologies to save iron from corrosion.</p> <p>Steel industry:</p> <p>Steel – Mechanical properties of materials and change with respect to temperature, phase diagram of Fe-C system, manufacturing of steel, classification of steel, heat treatment of steel, important alloys of iron and their applications. Types of Corrosion and passivation techniques</p> <p>Metal finishing technology:</p> <p>Introduction, need for surface treatment, different surface finishing processes, basics of electrodeposition, electroplating principles, electrochemistry applied to electroplating, mechanical preparation of surfaces - pickling, cleaning, rinsing, composition and conditions of plating bath, electroplating of metals-chromium, nickel, electroplating of plastics, electroplating waste treatment and metal recovery.</p>					
Learning Outcomes					
<p>On the completion of the course:</p> <ul style="list-style-type: none"><li>• Students are expected to become familiarized with the concepts of Steel Metallurgy and Electroplating</li><li>• This will enable them qualify for basic to moderate level jobs involving general knowledge of chemistry</li><li>• The obtained knowledge shall also enable the students to enter into various entrepreneurial activities involving general introduction to chemistry</li><li>• Students are able to understand the concept of GLP and GMP</li></ul>					
Course Content			Assignments/Readings		
Week 1	Mechanical properties of materials and change with respect to temperature		Class Based learning/tests		
Week 2	Mechanical properties of materials and change with respect to temperature, phase diagram of Fe-C system		Class Based learning/tests		
Week 3	Manufacturing of Steel		Class Based learning/tests		
Week 4	Manufacturing of Steel		Class Based learning/tests		
Week 5	Important alloys of iron and their applications		Class Based learning/tests		
Week 6	Types of Corrosion and passivation techniques		Class Based learning/tests		
Week 7	Types of Corrosion and passivation techniques		Written Assignment		
Week 8	Heat Treatments of Steel		Class Based learning/tests		
Week 9	Mid-term Assessment		----		
Week 10	Introduction of Metal Finishing Technology		Class Based learning/tests		
Week 11	Need of Surface treatment		Class Based learning/tests		
Week 12	Different surface finishing processes		Class Based learning/tests		

<b>Week 13</b>	Basics of electrodeposition, electroplating principles	Class Based learning/tests
<b>Week 14</b>	Electrochemistry applied to electroplating, mechanical preparation of surfaces - pickling, cleaning, rinsing, composition	Class Based learning/tests
<b>Week 15</b>	Conditions of plating bath, electroplating of metals-chromium, nickel	Quiz
<b>Week 16</b>	Electroplating of plastics, electroplating waste treatment and metal recovery	Class Based learning/tests

#### **Textbooks and Reading Material**

1. Applied Chemistry, Haq Nawaz Bhatti and Muhammad Salman, 2017, Caravan Book Publisher, Pakistan.
2. Chemistry of iron and Steel Manufacture, C. Bodsworth, Longman Press, London, 1963.
3. Graham's Electroplating Engineering Hand Book, Ed. L.J. Durney, CBS Publishers and Distributors, New Delhi. (1997).
4. Nickel and Chromium plating, J.K. Dennis & T.E. Such, Newness Butterworth, London (1972).

#### **Teaching Learning Strategies**

1. Lectures
2. Group Discussion
3. Laboratory work
4. Seminar/ Workshop

#### **Assignments: Types and Number with Calendar**

1. Written 7<sup>th</sup> week
2. Quiz 15<sup>th</sup> week

#### **Assessment**

<b>Sr. No.</b>	<b>Elements</b>	<b>Weightage</b>	<b>Details</b>
1.	Midterm Assessment	35%	Written Assessment at the mid-point of the semester.
2.	Formative Assessment	25%	Continuous assessment includes: Classroom participation, assignments, presentations, viva voce, attitude and behavior, hands-on-activities, short tests, projects, practical, reflections, readings, quizzes etc.
3.	Final Assessment	40%	Written Examination at the end of the semester. It is mostly in the form of a test, but owing to the nature of the course the teacher may assess their students based on term paper, research proposal development, field work and report writing etc.

Semester VII					
Programme	BS Chemistry	Course Code	Chem-464	Credit Hours	1
Course Title	Applied Chemistry Lab-II		Course Title	Major (Elective)	
Course Introduction					
Students will learn how to apply basic and instrumental techniques to determine metal ions in metallurgical samples. Analysis Iron in steel by titrimetry, Analysis of Nickel in steel by Gravimetry and Solvent Extraction, Analysis of Chromium in steel by Spectrophotometry, Heavy metal analysis of steel by AAS, Analysis of dolomite, chromite and bauxite Ore by titration method.					
Learning Outcomes					
On the completion of the course: <ul style="list-style-type: none"><li>Students are expected to get familiarized with the concepts of general chemistry</li><li>This will enable them qualify for basic to moderate level jobs involving general knowledge of chemistry</li><li>The obtained knowledge shall also enable the students to enter into various entrepreneurial activities involving general introduction to chemistry</li><li>Students are able to understand the concept of GLP and GMP</li></ul>					
Course Content			Assignments/Readings		
Week 1	Analysis of Iron in steel by titrimetry		Lab work / Notebook		
Week 2	Analysis of Iron in steel by titrimetry		Lab work / Notebook		
Week 3	Analysis of Nickel in steel by Gravimetry		Lab work / Notebook		
Week 4	Analysis of Nickel in steel by Solvent Extraction		Lab work / Notebook		
Week 5	Analysis of Nickel in steel by Solvent Extraction		Lab work / Notebook		
Week 6	Analysis of Chromium in steel by spectrophotometry		Lab work / Notebook		
Week 7	Analysis of Chromium in steel by spectrophotometry		Written Assignment		
Week 8	Heavy metals analysis of steel by AAS		Lab work / Notebook		
Week 9	Mid Term Examination		Lab work / Notebook		
Week 10	Heavy metals analysis of steel by AAS		Lab work / Notebook		
Week 11	Heavy metals analysis of steel by AAS		Lab work / Notebook		
Week 12	Analysis of Dolomite Ore		Lab work / Notebook		
Week 13	Analysis of Chromite Ore		Lab work / Notebook		
Week 14	Analysis of Chromite Ore		Lab work / Notebook		
Week 15	Analysis of Bauxite Ore		Quiz		
Week 16	Review and Viva Voce		Lab work / Notebook		

Textbooks and Reading Material			
1. Applied Chemistry, Haq Nawaz Bhatti and Muhammad Salman, 2017, Caravan Book Publisher, Pakistan. 2. Chemistry of iron and Steel Manufacture, C. Bodsworth, Longman Press, London, 1963. 3. Graham's Electroplating Engineering Hand Book, Ed. L.J. Durney, CBS Publishers and Distributors, New Delhi. (1997). 4. Nickel and Chromium plating, J.K. Dennis & T.E. Such, Newness Butterworth, London (1972). 5. Applied Chemistry by Vermani			
Teaching Learning Strategies			
1. Lectures 2. Group Discussion 3. Laboratory work 4. Seminar/ Workshop			
Assignments: Types and Number with Calendar			
1. Written 7 <sup>th</sup> week 2. Quiz 15 <sup>th</sup> week			
Assessment			
Sr. No.	Elements	Weightage	Details
1.	Midterm Assessment	35%	Written Assessment at the mid-point of the semester.
2.	Formative Assessment	25%	Continuous assessment includes: Classroom participation, assignments, presentations, viva voce, attitude and behavior, hands-on-activities, short tests, projects, practical, reflections, readings, quizzes etc.
3.	Final Assessment	40%	Written Examination at the end of the semester. It is mostly in the form of a test, but owing to the nature of the course the teacher may assess their students based on term paper, research proposal development, field work and report writing etc.



Semester VII					
Programme	BS Chemistry	Course Code	Chem-465	Credit Hours	2
Course Title	Separation and Spectral Techniques		Course Title	Major (Elective)	
Course Introduction					
<p>Students will learn about the principle, theory and working of basic analytical techniques.</p> <p>Chromatography</p> <p>Thin Layer Chromatography – Basic principle, theory and mechanism, stationary and mobile phase selection, locating reagents, applications, introduction to HPTLC</p> <p>High Performance Liquid Chromatography (HPLC) – Basic principle, types, theory and mechanism, stationary and mobile phase selection and types, basic parts of HPLC including pump, column, injector, detector, thermostat etc. Explanation of typical chromatogram highlighting retention time, peak height and width, tailing factor, resolution, theoretical plates, Isocratic and gradient elution and its significance, HPLC detectors such as Refractive Index, UV/Vis, photodiode array and fluorescence detector, Applications of HPLC</p> <p>Gas Chromatography Mass Spectrometry (GCMS) – Basic principle, theory and mechanism, stationary and mobile phase selection and types, basic parts of typical gas chromatography including sample injection port, gas reservoir, column and detectors. Nature of samples to be analyzed by GCMS, temperature selection, packed and capillary columns, Mass analyzers, quaderpole mass analyzers, Time of flight analyzer Applications of GCMS</p> <p>Spectroscopy</p> <p>Introduction, Basic Principle, Theory and Applications of Flame emission spectroscopy (FES), Flame atomic absorption spectroscopy (FAAS), continuous and line sources, construction of hollow cathode lamp, types of flames and their appropriate use, sensitivity and detection limits, Optical, Chemical, Physical and Ionization interferences and their possible solutions, Introduction, Basic Principle, Theory and Applications of UV/Vis spectroscopy and Thermal Analysis Techniques.</p>					
Learning Outcomes					
<p>On the completion of the course:</p> <ul style="list-style-type: none"><li>• Students are expected to become familiarized with the concepts of basic Analytical Techniques</li><li>• This will enable them qualify for basic to moderate level jobs involving general knowledge of Chemistry</li><li>• The obtained knowledge shall also enable the students to enter into various entrepreneurial activities involving general introduction to chemistry</li><li>• Students are able to understand the concept of GLP and GMP</li></ul>					
Course Content			Assignments/Readings		
Week 1	Thin Layer Chromatography – Basic principle, theory and mechanism, stationary and mobile phase selection, locating reagents, applications, introduction to HPTLC		Class Based learning/tests		
Week 2	Thin Layer Chromatography – Basic principle, theory and mechanism, stationary and mobile phase selection, locating reagents, applications, introduction to HPTLC		Class Based learning/tests		

<b>Week 3</b>	High Performance Liquid Chromatography (HPLC) – Basic principle, types, theory and mechanism, stationary and mobile phase selection and types,	Class Based learning/tests
<b>Week 4</b>	Basic parts of HPLC including pump, column, injector, detector, thermostat.	Class Based learning/tests
<b>Week 5</b>	Explanation of typical chromatogram highlighting retention time, peak height and width, tailing factor, resolution, theoretical plates, Isocratic and gradient elution and its significance.	Class Based learning/tests
<b>Week 6</b>	Explanation of typical chromatogram highlighting retention time, peak height and width, tailing factor, resolution, theoretical plates, Isocratic and gradient elution and its significance	Class Based learning/tests
<b>Week 7</b>	HPLC detectors such as Refractive Index, UV/Vis, photodiode array and fluorescence detector, Applications of HPLC	Written Assignment
<b>Week 8</b>	Gas Chromatography Mass Spectrometry (GCMS) – Basic principle, theory and mechanism, stationary and mobile phase selection and types,	Class Based learning/tests
<b>Week 9</b>	Midterm Assessment	----
<b>Week 10</b>	Basic parts of typical gas chromatography including sample injection port, gas reservoir, column and detectors.	Class Based learning/tests
<b>Week 11</b>	Nature of samples to be analyzed by GCMS, temperature selection,	Class Based learning/tests
<b>Week 12</b>	Packed and capillary columns, Mass analyzers, quadrupole mass analyzers, Time of flight analyzer, Applications of GCMS	Class Based learning/tests
<b>Week 13</b>	Introduction, Basic Principle, Theory and Applications of Flame Emission Spectroscopy (FES),	Class Based learning/tests
<b>Week 14</b>	Flame atomic absorption spectroscopy (FAAS), continuous and line sources, construction of hollow cathode lamp, types of flames and their appropriate use, sensitivity and detection limits. Optical, Chemical, Physical and Ionization interferences and their possible solutions.	Class Based learning/tests
<b>Week 15</b>	Flame atomic absorption spectroscopy (FAAS), continuous and line sources, construction of hollow cathode lamp, types of flames and their appropriate use, sensitivity and detection limits, Optical, Chemical, Physical and Ionization interferences and their possible solutions,	Quiz
<b>Week 16</b>	Introduction, Basic Principle, Theory and Applications of UV/Vis spectroscopy and Thermal Analysis Techniques.	Class Based learning/tests

Textbooks and Reading Material			
<ol style="list-style-type: none"> <li>1. T. B. of Quantitative Inorganic Analysis, Vogel's Ed-4<sup>th</sup>, Longman Group Limited (1978).</li> <li>2. Applied Chemistry, Haq Nawaz Bhatti and Muhammad Salman, 2017, Caravan Book Publisher, Pakistan.</li> <li>3. Instrumental Analysis, Gary D. Christain, 1978, Introduction to Instrumental Analysis by Braun, McGraw-Hill Book company, 1987.</li> <li>4. Instrumental Analysis by B.K. Sharma</li> </ol>			
Teaching Learning Strategies			
<ol style="list-style-type: none"> <li>1. Lectures</li> <li>2. Group Discussion</li> <li>3. Laboratory work</li> <li>4. Seminar/ Workshop</li> </ol>			
Assignments: Types and Number with Calendar			
<ol style="list-style-type: none"> <li>1. Written 7<sup>th</sup> week</li> <li>2. Quiz 15<sup>th</sup> week</li> </ol>			
Assessment			
Sr. No.	Elements	Weightage	Details
1.	Midterm Assessment	35%	Written Assessment at the mid-point of the semester.
2.	Formative Assessment	25%	Continuous assessment includes: Classroom participation, assignments, presentations, viva voce, attitude and behavior, hands-on-activities, short tests, projects, practical, reflections, readings, quizzes etc.
3.	Final Assessment	40%	Written Examination at the end of the semester. It is mostly in the form of a test, but owing to the nature of the course the teacher may assess their students based on term paper, research proposal development, field work and report writing etc.

Semester VII					
Programme	BS Chemistry	Course Code	Chem-466	Credit Hours	1
Course Title	Applied Chemistry Lab-III		Course Title	Major (Elective)	
Course Introduction					
Student will apply the analytical techniques for characterization of Industrial samples. Spectrophotometric analysis of iron in pharmaceutical tablets, Chlorides in water, phosphates in fertilizers, Chromium in Tannery wastewater, Recovery of chromium from tannery waste water, Application of AAS on analysis of heavy metals of various industrial effluents.					
Learning Outcomes					
On the completion of the course: <ul style="list-style-type: none"><li>• Students are expected to become familiarized with the concepts of general chemistry</li><li>• This will enable them qualify for basic to moderate level jobs involving general knowledge of Chemistry</li><li>• The obtained knowledge shall also enable the students to enter into various entrepreneurial activities involving general introduction to chemistry</li><li>• Students are able to understand the concept of GLP and GMP</li></ul>					
Course Content				Assignments/Readings	
Week 1	Spectrophotometric analysis of iron in pharmaceutical tablets			Lab work / Notebook	
Week 2	Spectrophotometric analysis of iron in pharmaceutical tablets			Lab work / Notebook	
Week 3	Determination of Chlorides in water by spectrophotometry			Lab work / Notebook	
Week 4	Determination of Chlorides in water by spectrophotometry			Lab work / Notebook	
Week 5	Determination of phosphates in Fertilizers			Lab work / Notebook	
Week 6	Determination of phosphates in Fertilizers			Lab work / Notebook	
Week 7	Analysis of Chromium in Tannery wastes			Written Assignment	
Week 8	Analysis of Chromium in Tannery wastes			Lab work / Notebook	
Week 9	Mid Term Examination			Lab work / Notebook	
Week 10	Recovery of Chromium from Tannery wastes			Lab work / Notebook	
Week 11	Application of AAS on analysis of heavy metals of various industrial effluents.			Lab work / Notebook	
Week 12	Application of AAS on analysis of heavy metals of various industrial effluents.			Lab work / Notebook	
Week 13	Application of AAS on analysis of heavy metals of various industrial effluents.			Lab work / Notebook	
Week 14	Application of AAS on analysis of heavy metals of various industrial effluents.			Lab work / Notebook	

<b>Week 15</b>	Application of AAS on analysis of heavy metals of various industrial effluents.	Quiz	
<b>Week 16</b>	Review and Viva Voce	Lab work / Notebook	
<b>Textbooks and Reading Material</b>			
1. T. B. of Quantitative Inorganic Analysis, Vogel’s Ed-4 <sup>th</sup> , Longman Group Limited (1978). 2. Applied Chemistry, Haq Nawaz Bhatti and Muhammad Salman, 2017, Caravan Book Publisher, Pakistan. 3. Instrumental Analysis, Gary D. Christain, 1978, Introduction to Instrumental Analysis by Braun, McGraw-Hill Book company, 1987. <b>4. Instrumental Analysis by B.K. Sharma</b>			
<b>Teaching Learning Strategies</b>			
1. Lectures 2. Group Discussion 3. Laboratory work 4. Seminar/ Workshop			
<b>Assignments: Types and Number with Calendar</b>			
1.Written 7 <sup>th</sup> week 2.Quiz 15 <sup>th</sup> week			
<b>Assessment</b>			
<b>Sr. No.</b>	<b>Elements</b>	<b>Weightage</b>	<b>Details</b>
1.	Midterm Assessment	35%	Written Assessment at the mid-point of the semester.
2.	Formative Assessment	25%	Continuous assessment includes: Classroom participation, assignments, presentations, viva voce, attitude and behavior, hands-on-activities, short tests, projects, practical, reflections, readings, quizzes etc.
3.	Final Assessment	40%	Written Examination at the end of the semester. It is mostly in the form of a test, but owing to the nature of the course the teacher may assess their students based on term paper, research proposal development, field work and report writing etc.

Semester VII					
Programme	BS Chemistry	Course Code	Chem-467	Credit Hours	3
Course Title	Processing Industries		Course Title	Major (Elective)	
Course Introduction					
<p>Students will understand and learn about the operation and processes involved in Paper, Leather and Oils/Fats industries starting from the raw material(s) to the end product.</p> <p>Paper &amp; leather industry</p> <p>Paper – History, Survey of Raw materials, Production of Pulp by Soda process, Sulphite process, Kraft (Sulphate) process, Recovery of cooking liquor, manufacturing of paper, Environmental aspects of paper Industry.</p> <p>Leather – Introduction, Types of Skin, Theory of Tanning, Beamhouse Operations, Vegetable and Chrome tanning, Finishing processes, Waste Disposal and Pollution Aspects.</p> <p>Oils and fats</p> <p>Classification of oils and fats, vegetable oils, essential oils, various methods of extraction of oils, refining and hydrogenation of oils, Industrial applications of oils in resins, surfactants, lubricants and paints.</p>					
Learning Outcomes					
<p>On the completion of the course, the students will:</p> <ul style="list-style-type: none"><li>• Students are expected to get familiarized with the concepts of basic Analytical Techniques</li><li>• This will enable them qualify for basic to moderate level jobs involving general knowledge of Chemistry.</li><li>• The obtained knowledge shall also enable the students to enter into various entrepreneurial activities involving general introduction to chemistry.</li></ul>					
Course Content				Assignments/Readings	
Week 1	Paper – History, Survey of Raw materials			Class Based learning/tests	
Week 2	Production of Pulp by Soda process			Class Based learning/tests	
Week 3	Production of Pulp by Sulphite process			Class Based learning/tests	
Week 4	Production of Pulp by Kraft (Sulphate) process			Class Based learning/tests	
Week 5	Recovery of cooking liquor			Class Based learning/tests	
Week 6	Manufacturing of paper			Class Based learning/tests	
Week 7	Environmental aspects of the Paper Industry			Written Assignment	
Week 8	Leather – Introduction, Types of Skin, Theory of Tanning			Class Based learning/tests	
Week 9	Midterm Assessment			----	
Week 10	Beamhouse Operations			Class Based learning/tests	
Week 11	Vegetable and Chrome tanning			Class Based learning/tests	
Week 12	Finishing processes, Waste Disposal and Pollution Aspects.			Class Based learning/tests	

<b>Week 13</b>	Classification of oils and fats, vegetable oils, essential oils	Class Based learning/tests
<b>Week 14</b>	Various methods of extraction of oils	Class Based learning/tests
<b>Week 15</b>	Refining and hydrogenation of oils	Quiz
<b>Week 16</b>	Industrial applications of oils in resins, surfactants, lubricants and paints	Class Based learning/tests

### Textbooks and Reading Material

1. Applied Chemistry, Haq Nawaz Bhatti and Muhammad Salman, 2017, Caravan Book Publisher, Pakistan.
2. Pulp and Paper Technology, Testing and Applications, K.P. Rao (2003), CBS Publishers.
3. Chemistry of Pulp and Paper making, Edwin Sutermeister, Ed-3<sup>rd</sup> (1946)
4. Fertilizers and Soil Fertility, U.S.Jones, Reston Publishing Co. Virginia, 1979.
5. Industrial chemistry, B. K. Sharma, Krishna Prakashan Media (P) Ltd., Ed-15 (2006).
6. Shereve's Chemical Process Industries, 5th Ed.1975, by G.T.Austin, McGraw Hill Book Co. New York.

### Teaching Learning Strategies

1. Lectures
2. Group Discussion
3. Laboratory work
4. Seminar/ Workshop

### Assignments: Types and Number with Calendar

1. Written 7<sup>th</sup> week
2. Quiz 15<sup>th</sup> week

### Assessment

Sr. No.	Elements	Weightage	Details
1.	Midterm Assessment	35%	Written Assessment at the mid-point of the semester.
2.	Formative Assessment	25%	Continuous assessment includes: Classroom participation, assignments, presentations, viva voce, attitude and behavior, hands-on-activities, short tests, projects, practical, reflections, readings, quizzes etc.
3.	Final Assessment	40%	Written Examination at the end of the semester. It is mostly in the form of a test, but owing to the nature of the course the teacher may assess their students based on term paper, research proposal development, field work and report writing etc.